

## Uranium oxide interference correction algorithms

Measurement of the uranium isotopic composition of double-spiked ( $^{233}\text{U}$ - $^{235}\text{U}$ ) samples as  $\text{UO}_2^+$  species requires the subtraction of the isobaric interference of  $^{18}\text{O}$ -bearing  $^{233}\text{UO}_2$  on the predominant target  $^{235}\text{U}^{16}\text{O}_2$  peak at mass 267. The probability of forming a double  $^{17}\text{O}$ -bearing uranium dioxide molecule is so low as to make a  $^{233}\text{U}^{17}\text{O}_2$  interference of negligible importance for this measurement. The probability of an  $^{18}\text{O}$  dioxide substitution is estimated by the relative abundance of  $^{18}\text{O}$ , also safely approximated by  $^{18}\text{O}/^{16}\text{O}$ , which is assigned a value of 0.002 (IUPAC, 1997). Both possible  $^{18}\text{O}$  substitutions into the dioxide molecule must be considered. The maths for the correction of the two measured U isotope ratios are given below.

For correcting the  $^{235}\text{U}/^{233}\text{U}$  (or  $^{233}\text{U}/^{235}\text{U}$ ) ratio:

$$\left(\frac{267}{265}\right)_{meas} = \frac{^{235}\text{U}^{16}\text{O}^{16}\text{O} + ^{233}\text{U}^{18}\text{O}^{16}\text{O} + ^{233}\text{U}^{16}\text{O}^{18}\text{O}}{^{233}\text{U}^{16}\text{O}^{16}\text{O}} \quad (1)$$

$$^{233}\text{U}^{18}\text{O}^{16}\text{O} = ^{233}\text{U}^{16}\text{O}^{18}\text{O} = ^{233}\text{U}^{16}\text{O}^{16}\text{O} \left(\frac{^{18}\text{O}}{^{16}\text{O}}\right) \quad (2)$$

$$\left(\frac{267}{265}\right)_{meas} = \frac{^{235}\text{U}^{16}\text{O}^{16}\text{O} + ^{233}\text{U}^{16}\text{O}^{16}\text{O} \left(\frac{^{18}\text{O}}{^{16}\text{O}}\right) + ^{233}\text{U}^{16}\text{O}^{16}\text{O} \left(\frac{^{18}\text{O}}{^{16}\text{O}}\right)}{^{233}\text{U}^{16}\text{O}^{16}\text{O}} \quad (3)$$

$$\left(\frac{267}{265}\right)_{meas} = \frac{^{235}\text{U}^{16}\text{O}^{16}\text{O} + ^{233}\text{U}^{16}\text{O}^{16}\text{O} \left[ \left(\frac{^{18}\text{O}}{^{16}\text{O}}\right) + \left(\frac{^{18}\text{O}}{^{16}\text{O}}\right) \right]}{^{233}\text{U}^{16}\text{O}^{16}\text{O}} \quad (4)$$

$$\left(\frac{267}{265}\right)_{meas} = \frac{^{235}\text{U}^{16}\text{O}^{16}\text{O}}{^{233}\text{U}^{16}\text{O}^{16}\text{O}} + \left[ \left(\frac{^{18}\text{O}}{^{16}\text{O}}\right) + \left(\frac{^{18}\text{O}}{^{16}\text{O}}\right) \right] \quad (5)$$

$$\frac{^{235}\text{U}^{16}\text{O}^{16}\text{O}}{^{233}\text{U}^{16}\text{O}^{16}\text{O}} = \left(\frac{267}{265}\right)_{true} \quad (6)$$

$$\left(\frac{267}{265}\right)_{meas} = \left(\frac{267}{265}\right)_{true} + \left[ \left(\frac{^{18}\text{O}}{^{16}\text{O}}\right) + \left(\frac{^{18}\text{O}}{^{16}\text{O}}\right) \right] \quad (7)$$

$$\left(\frac{^{18}\text{O}}{^{16}\text{O}}\right) = 0.002 \quad (\text{IUPAC 1997}) \quad (8)$$

$$\left(\frac{267}{265}\right)_{meas} = \left(\frac{267}{265}\right)_{true} + [0.002 + 0.002] = \left(\frac{267}{265}\right)_{true} + 0.004 \quad (9)$$

$$\left(\frac{267}{265}\right)_{true} = \left(\frac{267}{265}\right)_{meas} - 0.004 \quad (10)$$

$$\left(\frac{^{233}U}{^{235}U}\right) = \left(\frac{265}{267}\right)_{true} = \frac{1}{\left(\frac{267}{265}\right)_{true}} = \frac{1}{\left(\frac{267}{265}\right)_{meas} - 0.004} = \frac{1}{\left(\frac{265}{267}\right)_{meas} - 0.004} \quad (11)$$

The final form of eqn. 11 accommodates those accustomed to both measuring and then entering the  $^{233}U/^{235}U$  ratio into data reduction software.

For correcting the  $^{238}U/^{235}U$  ratio:

$$\left(\frac{270}{267}\right)_{true} = \left(\frac{270}{265}\right)_{meas} \times \left(\frac{265}{267}\right)_{true} \quad (12)$$

$$\left(\frac{270}{267}\right)_{true} = \left(\frac{270}{265}\right)_{meas} \times \left(\frac{265}{267}\right)_{true} \quad (13)$$

Combining with eqn. 11...

$$\left(\frac{^{238}U}{^{235}U}\right) = \left(\frac{270}{267}\right)_{true} = \left(\frac{270}{265}\right)_{meas} \times \frac{1}{\left(\frac{265}{267}\right)_{meas} - 0.004} \quad (14)$$

For those used to measuring the  $^{238}U/^{235}U$  and  $^{233}U/^{235}U$  ratios...

$$\left(\frac{270}{265}\right)_{meas} = \frac{\left(\frac{270}{267}\right)_{meas}}{\left(\frac{265}{267}\right)_{meas}} \quad (15)$$

$$\left(\frac{^{238}U}{^{235}U}\right) = \left(\frac{270}{267}\right)_{true} = \frac{\left(\frac{270}{267}\right)_{meas} / \left(\frac{265}{267}\right)_{meas}}{\left(\frac{265}{267}\right)_{meas} - 0.004} \quad (16)$$

This last form accommodates those accustomed to both measuring and entering the  $^{238}\text{U}/^{235}\text{U}$  and  $^{233}\text{U}/^{235}\text{U}$  ratios into data reduction software. For Sector-54 users measuring on Faraday cups, the corrections (eqns. 11 and 16) can be simply added as functions to static multi-collection ADS files set up in parallel with existing U metal files but measuring masses 265, 267, and 270. For single collector Daly measurements, these corrections can be done off-line (for example as a group of cells within the 'Reduce Data' page of Pb-MacDat).

These algorithms, including the assumed  $^{18}\text{O}$  abundance were confirmed empirically at MIT by splitting large spiked sample U loads and running as both metal and oxide. When combined with the sample Pb isotope abundances, there was excellent agreement (better than 0.1%) in calculated Pb/U ratios between the metal and corrected oxide uranium isotope measurements.

Mark Schmitz  
November 3, 2004